

# Depleted Zinc-64 in the form of Zinc Oxide and Zinc Acetate Dihydrate



- Zinc, depleted in the isotope  $^{64}\text{Zn}$ , is a product for the water chemistry of nuclear power plants (NPPs).
- Depleted Zinc in the form of sintered Zinc Oxide Pellets is used in boiling water reactors (BWR).
- Depleted Zinc in the form of Zinc Oxide Powder or Zinc Acetate is used in pressurized water reactors (PWR).
- The aim of using Depleted Zinc is to reduce corrosion in the primary water circuit in order to reduce occupational radiation exposure for NPP personnel (possible reduction of collective dose rate is approx. 50 %).
- Another goal is the reduction of the potential for stress corrosion cracking in PWRs.
- American BWRs (mainly GE types) started with the use of Zinc Oxide in the form of natural zinc. The residual dose uptake problem from the activation of  $^{64}\text{Zn}$  to  $^{65}\text{Zn}$  has been successfully avoided in the United States, Mexico and some European countries by using Depleted Zinc.

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# Zinc Oxide Pellets

## Specification

### Physical Properties:

- Material  $^{64}\text{Zn}$  Depleted Zinc Oxide in form of sintered pellets
- Depletion  $^{64}\text{Zn} \leq 1 \text{ at}\%$
- Density  $\geq 4,77 \text{ g/cm}^3$
- Diameter  $10 \text{ mm} \pm 1 \text{ mm}$
- Length  $10 \text{ mm} \pm 1 \text{ mm}$

### Chemical Properties:

- Form ZnO
- Purity  $\geq 99,8 \text{ wt}\%$

## Impurities in $\mu\text{g/g}$

Ag	$\leq 10$
Al	$\leq 50$
As, Au	$\leq 10$
B, Ba, Be, Bi, Br	$\leq 10$
C	$\leq 500$
Ca, Cd, Ce, Cl, Co, Cr, Cs, Cu	$\leq 10$
Dy	$\leq 10$
Er, Eu	$\leq 10$
F, Fe	$\leq 10$
Ga, Gd, Ge	$\leq 10$
Hf	$\leq 10$
Hg	$\leq 2$
Ho	$\leq 10$
I, In, Ir	$\leq 10$
K	$\leq 10$
La, Li, Lu	$\leq 10$
Mg, Mn, Mo	$\leq 10$
N, Na, Nb, Nd, Ni	$\leq 10$
Os	$\leq 10$
P, Pb, Pd, PO <sub>4</sub> , Pr, Pt	$\leq 10$
Rb, Re, Rh, Ru	$\leq 10$
S, Sb, Sc, Se	$\leq 10$
Si	$\leq 50$
Sn	$\leq 50$
Sm, SO <sub>4</sub> , Sr	$\leq 10$
Ta, Tb, Te, Th, Ti, Tl, Tm	$\leq 10$
U	$\leq 10$
V	$\leq 10$
W	$\leq 10$
Y, Yb	$\leq 10$
Zr	$\leq 10$
Rare Earth	$\leq 20$
Insoluble	$\leq 100$
Remaining Impurities	$\leq 20$

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# Zinc Acetate Dihydrate Powder

## Specification

### Physical Properties:

- Material  $^{64}\text{Zn}$  Depleted Zinc in form of Zinc Acetate Dihydrate
- Zn-64  $^{64}\text{Zn} \leq 1 \text{ at}\%$
- Particle Size  $\leq 1 \text{ mm}$

### Chemical Properties:

- Form  $\text{Zn}(\text{CH}_3\text{CO}_2)_2(\text{H}_2\text{O})_2$
- Purity  $\geq 99,8 \text{ wt}\%$

## Impurities in $\mu\text{g/g}$

Ag	$\leq 10$
Al	$\leq 50$
As	$\leq 10$
Au	$\leq 20$
B, Ba	$\leq 10$
Be	$\leq 20$
Bi, Br	$\leq 10$
C	n.s.
Ca, Cd	$\leq 10$
Ce, Cl	$\leq 20$
Co, Cr	$\leq 10$
Cs	$\leq 20$
Cu	$\leq 10$
Dy	$\leq 20$
Er, Eu	$\leq 20$
F	$\leq 20$
Fe	$\leq 10$
Ga, Gd, Ge	$\leq 20$
Hf	$\leq 10$
Hg, Ho	$\leq 20$
I, In	$\leq 10$
Ir	$\leq 20$
K	$\leq 10$
La	$\leq 20$
Li	$\leq 10$
Lu	$\leq 20$
Mg, Mn, Mo	$\leq 10$
N	$\leq 20$
Na	$\leq 10$
Nb, Nd	$\leq 20$
Ni	$\leq 10$
Os	$\leq 20$
P, Pb	$\leq 10$

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# Zinc Acetate Dihydrate Powder

Pd	≤ 20
PO <sub>4</sub> , Pr, Pt	≤ 20
Rb, Re, Rh, Ru	≤ 20
S, Sb	≤ 10
Sc	≤ 20
Se	≤ 10
Si	≤ 50
Sm	≤ 20
Sn	≤ 10
SO <sub>4</sub>	≤ 50
Sr	≤ 10
Ta, Tb, Te, Th	≤ 20
Ti	≤ 10
Tl, Tm	≤ 20
U	≤ 20
V	≤ 10
W	≤ 20
Y, Yb	≤ 20
Zr	≤ 20
Rare Earth	≤ 20
Insoluble	≤ 100
Remaining Impurities	≤ 10

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